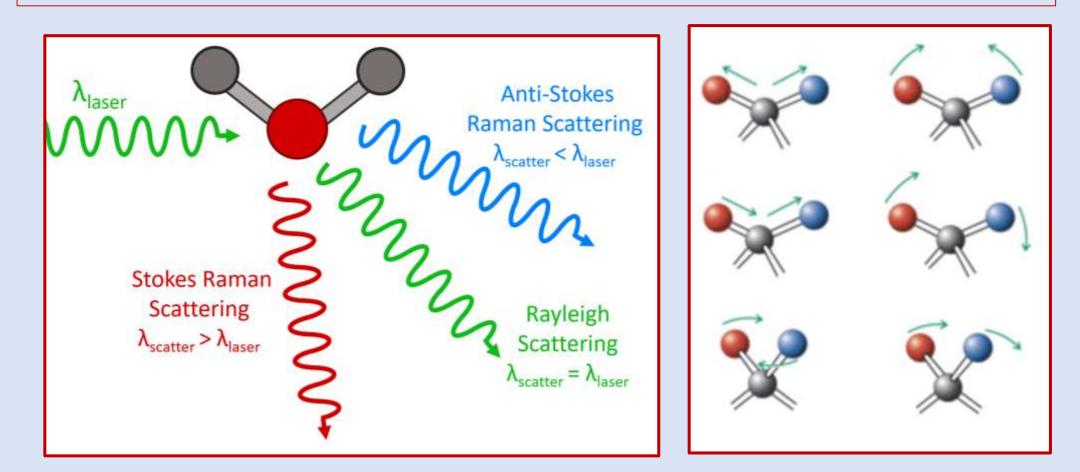


Introduction to Raman Spectroscopy

Raman spectroscopy is based on measurements of the inelastic scattering of monochromatic light from molecules. During the scattering process, the interactions of incident photons with bonding electrons of the analyte result in different, up or down, energy changes of photons, due to the exchange of energy with molecular vibrational modes.

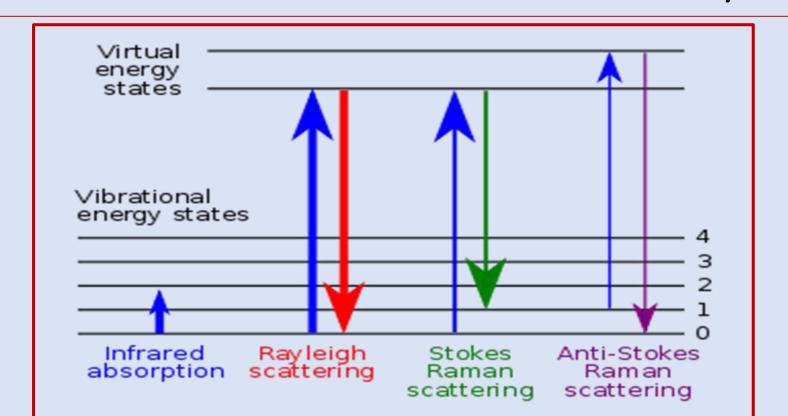
The light scattering spectrum represents the observed intensities of these interactions versus changed energy, represented by the light frequency. Hence, the Raman spectrum is a "fingerprint" of intramolecular vibrations whose frequency is specific for the functional groups and bonds in the molecule and thus provides the means for their identification. Therefore, Raman spectroscopy can be utilized for the identification of molecules and determination of their concentration.

Raman scattering method is unique due to its ability to detect specific molecular vibrations with very high sensitivity, approaching 1 fM (1x10⁻¹⁵ mole/L) when used in Surface-Enhanced Raman Scattering (SERS) mode.

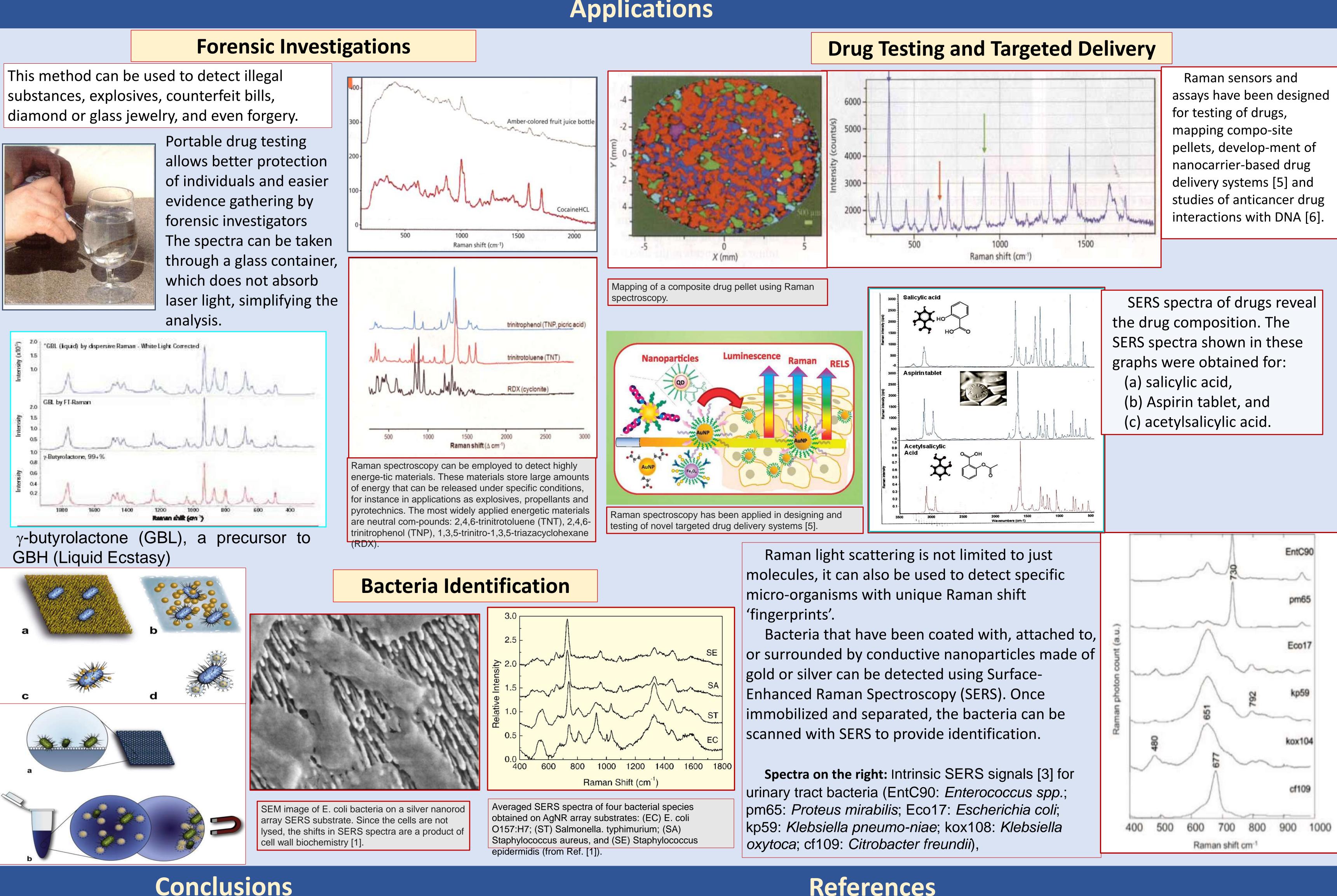


How does Raman Work?

Raman spectrometers utilize the Raman effect, in which the intramolecular vibration modes of the analyte absorb photons from the incident laser beam, followed by the re-emission of photons in all directions with a de-creased energy, leading to the so-called Stockes scatter-ing. The anti-Stockes scattering is also possible, when the re-emitted photons gain the energy from the excited vibration states. Since at the room temperature, most of the molecules are at the lowest energy level, stronger Raman signals are detected for the Stokes scattering. The Raman spectrum of the scattered light provides information on the molecular structure of the analyte.



Applications of Raman Spectroscopy in Science Lawrence M. Tucker, Lucas J. Scalcione, Maria Hepel Department of Chemistry, State University of New York, Potsdam, NY 13676



Raman spectroscopy is a rapid, highly sensitive, and effective tool for the identification of molecules and micro-organisms. Due to the widespread need for identification of materials and contaminant in industry, science, and medicine, it becomes one of the most sought after analytical techniques. Using the SERS effect, chemical compounds can be detected down to fM levels (1 in 10^{15} molecules).

Applications

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